



**ALDON  
CORPORATION**

## MATERIAL SAFETY DATA SHEET

1533 W.Henrietta Rd.  
Avon, New York 14414  
(716) 226-6177

MSDS No. BB 40  
Effective Date August 26, 1998

### SECTION I NAME 24 HOUR EMERGENCY ASSISTANCE

Product	BARIUM CHLORIDE, 0.1 NORMAL SOLUTION *
Chemical Synonyms	Barium Chloride Solution (0.1 N/ 0.05 M)
Formula	Mixture.
Unit Size	up to 4 Lt.
C.A.S. No.	Mixture.

CHEMTREC  
800-424-9300  
Day 716-226-6177

NFPA  
HAZARD RATING

HMIS \*

Health	2
Fire	0
Reactivity	0

LEAST SLIGHT MODERATE HIGH EXTREME  
0 1 2 3 4

### SECTION II INGREDIENTS OF MIXTURES

Principal Component(s)	%	TLV Units
Barium Chloride: (CAS No. 10361-37-2)	2.08%	See Section V.
Water: (CAS No. 7732-18-5)	97.92%	None established.

**WARNING! MAY BE HARMFUL OR FATAL IF SWALLOWED.**

**CONTACT CAUSES SKIN AND EYE IRRITATION.**

### SECTION III PHYSICAL DATA

Melting Point (°F)	Freezes approx. 0°C (32°F)	Specific Gravity (H <sub>2</sub> O = 1)	1.01
Boiling Point (°F)	100°C (212°F) water.	Percent Volatile by Volume (%)	97.92%
Vapor Pressure (mm Hg)	14 mm (water).	Evaporation Rate (Water = 1)	Slightly less than 1.
Vapor Density (Air=1)	0.7 (water).		
Solubility in Water	Complete.		
Appearance & Odor	Clear, colorless liquid; no odor.		

### SECTION IV FIRE AND EXPLOSION HAZARD DATA

Flash Point (Method Used)	Non-flammable.	Flammable Limits in Air % by Volume	N/A	Lower	Upper
Extinguisher Media	Use any media suitable for extinguishing supporting fire.				

#### SPECIAL FIREFIGHTING PROCEDURES

A NIOSH/MSHA-approved self-contained breathing apparatus should be worn for protection against barium-containing dust, mist or fumes.

#### UNUSUAL FIRE AND EXPLOSION HAZARDS

In fire conditions, water may evaporate from this solution, which may cause hazardous decomposition products to be produced as dust or fume.

D.O.T. NON-REGULATED

Approved by U.S. Department of Labor "essentially similar" to form OSHA-20

### SECTION V HEALTH HAZARD DATA

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Threshold Limited Value	0.5 mg/m <sup>3</sup> as Barium (ACGIH, 1992-93). U.S. H.E.W. Toxic Substance List (1976) gives; orl-hmu TDL: 80 mg/Kg.; orl-rat LDLo: 335 mg/Kg.
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Effects of Overexposure	<b>MAY BE HARMFUL OR FATAL IF SWALLOWED.</b> Oral intake may cause weakness, salivation and nausea, followed by vomiting and diarrhea. Patient may become cold and experience varying degrees of paralysis. <b>INHALATION:</b> May be harmful, with symptoms similar to those of oral intake. Observe allowable limits. <b>EYES AND SKIN CONTACT:</b> Not usually severe, but may cause irritation to sensitive individuals.
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Emergency and First Aid Procedures	<b>INGESTION:</b> If swallowed, if conscious, give one or two glasses of water to drink and induce vomiting. Repeat until vomit fluid is clear. Call physician immediately. Never give anything by mouth to an unconscious person. <b>EYES:</b> Flush thoroughly with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get medical attention. <b>SKIN:</b> Flush with water, then wash with mild soap and water. <b>INHALATION:</b> Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician.
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### SECTION VI REACTIVITY DATA

Stability	Unstable Stable	Conditions to Avoid	Excessive temperature and heat.
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Incompatibility (Materials to Avoid)	Bromine Trifluoride and 2-Furan percarboxylic acid, violent reaction.
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Hazardous Decomposition Products	Thermal decomposition or burning may produce hydrochloric acid, barium oxide and toxic fumes of chlorine gas.
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Hazardous Polymerization	Conditions to Avoid
May Occur	Will Not Occur
	X
	Not applicable.

### SECTION VII SPILL OR LEAK PROCEDURES

Steps to be taken in case material is released or spilled	In case of spill, treat the spill with dilute sulfuric acid solution to precipitate the barium to barium sulfate. Neutralize the excess acid using sodium bicarbonate. Flush residue to sewer with copious amounts of water.
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Waste Disposal Method	Discharge, treatment, or disposal may be subject to Federal, State or Local laws. These disposal guidelines are intended for the disposal of catalog-size quantities only.
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Dispose of by treating barium chloride solution with dilute sulfuric acid solution to precipitate the barium to barium sulfate. Neutralize the excess acid using sodium bicarbonate. Flush residue to sewer with copious amounts of water. Barium chloride solution may also be disposed of by contracting with a licensed waste disposal service.

### SECTION VIII SPECIAL PROTECTION INFORMATION

Respiration Protection (Specify Type)	None needed in normal laboratory handling. If misty conditions prevail, work in ventilation hood or wear a NIOSH/MSHA-approved respirator.
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Ventilation	Local Exhaust	None required.	Special	No.
	Mechanical (General)	None required.	Other	No.

Protective Gloves	Rubber.	Eye Protection	Chemical safety glasses.
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Other Protective Equipment	Goggles, smock, apron, proper gloves, eye wash station.
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### SECTION IX SPECIAL PRECAUTIONS

Precautions to be Taken in Handling & Storing	Store in a cool place, away from food products. Wash thoroughly after handling.
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Other Precautions	Read label on container before using. Do not wear contact lenses when working with chemicals.
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Avoid contact with skin, eyes and clothing. Avoid breathing mist. Remove and wash contaminated clothing.

For laboratory use only. Not for drug, food or household use. Keep out of reach of children.

Revision No. 6	Date 8/26/98	Approved Michael Raszeja	Chemical Safety Coordinator	MR
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