

IB *The Noble Metals*

In order of increasing atomic number the members of group IB of the periodic table are:

copper,
silver and
gold.

This group derives its name from the fact that these elements are often found in nature in elemental form. These elements are also called the "coinage" metals. The elements of this group are characterized by a single electron in an outer shell. Although copper and gold also have other valences, all of the elements in this group have a 1+ valence in common.